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Limonoids from the fruits of *Chisocheton lasiocarpus* (Meliaceae)

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ABSTRACT

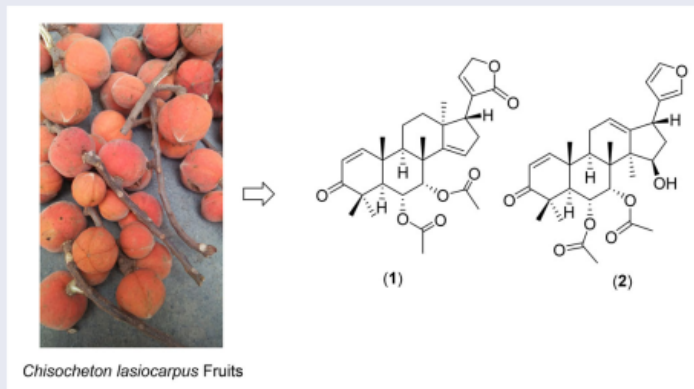
Two new azadirone-type limonoids, namely lasiocarpine A (**1**) and lasiocarpine B (**2**) were isolated from the fruit of *Chisocheton lasiocarpus* along with the known limonoids (**3–5**). UV, IR, one- and two-dimensional NMR, and mass spectrometry were used to determine the chemical structure of the isolated compounds. Furthermore, their cytotoxic activity against breast cancer cell line MCF-7 was evaluated using PrestoBlue reagent. From these compounds, lasiocarpine A (**1**) showed the strongest activity with an IC₅₀ value of 43.38 μ M.

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Chisocheton lasiocarpus Fruits

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1. Introduction

Limonoid is a class of naturally occurring compounds derived from triterpenoids that formed by the loss of four terminal carbons to form a 4,4,8-trimethyl-17-furanilsteroid skeleton. Further modification of the limonoid skeleton allows a wide and complex structural diversity of this compound [1–3]. They are produced abundantly in Meliaceae and Rutaceae plants but less common in Cneoraceae and Simaroubaceae families [4–6]. Limonoids from Meliaceae family have more diverse structures and one of the genera that contributes the most limonoid content is the *Chisocheton* genus. This type of compound has been reported to be found in all tissues including barks, stembarks, leaves, twigs, roots, seeds, and fruits [7, 8].

Chisocheton plants consist of 53 species distributed from eastern India, Nepal, Myanmar, South China, Thailand, Philippines, Malaysia, Papua New Guinea, and Indonesia to northern Australia [8, 9]. In Indonesia, these plants can be found on the islands of Sumatra, Java, Kalimantan, Sulawesi, and Papua [10, 11]. In the study conducted on limonoid compounds from Indonesian *Chisocheton* plants, dysobinol from *C. macrophyllus* seed [12] and pentandricine B-D from *C. pentandrus* stembark were reported [13]. In the further phytochemical investigation of *Chisocheton* plants, *C. lasiocarpus* was reported as an endemic plant in native habitat from Maluku to Solomon Islands [14]. However, there is still limited phytochemicals report of this plant. As continuous search for cytotoxic compound from Indonesian *Chisocheton* plants, five limonoids involving new azadirone-type limonoids (1–2) were isolated from the fruit of *C. lasiocarpus*. Herein, we report their isolation, structural elucidation and cytotoxic activity against MCF-7 breast cancer cells lines.

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2. Results and discussion

The methanol extract of *C. lasiocarpus* fruit was dissolved in water and extracted by liquid-liquid extraction techniques successively using hexane, ethyl acetate, and *n*-butanol. The phytochemical analysis of the extracts showed that the ethyl acetate extract contained an abundance of limonoid content. The separation of the ethyl acetate extract was conducted using column chromatography on silica gel and octadecyl silane as the stationary phase to give five limonoids, 1–5 (Figures 1 and S1).

Lasiocarpine A (1) was obtained as a colorless amorphous solid (MeOH) with a molecular formula of $C_{30}H_{38}O_7$ based on analysis of HR-TOFMS spectra with peak ion-molecule $[M+H]^+$ at m/z 511.2691 having twelve degrees of unsaturation. Meanwhile, the UV spectrum showed maximum absorption at 230 nm ($\log \epsilon$ 2.62), indicating the presence of a conjugated system. The IR spectrum showed the presence of C-H aliphatic (ν_{\max} 2917 cm^{-1}), conjugated carbonyl (ν_{\max} 1729 cm^{-1}), conjugated carbonyl ester (ν_{\max} 1676 cm^{-1}), *gem*-dimethyl (ν_{\max} 1447 cm^{-1}), and ether groups (ν_{\max} 1084 cm^{-1}). Furthermore, the $^1\text{H-NMR}$ spectrum (Table 1) showed five tertiary methyls typical for azadirone-type limonoid skeleton at δ_{H} 0.86 (3H, s, CH_3 -18), 1.25 (3H, s, CH_3 -19), 1.18 (3H, s, CH_3 -28), 1.34 (3H, s, CH_3 -30), 1.24 (3H, s, CH_3 -29), two additional methyl singlets from acetoxy groups at δ_{H} 2.00 (3H, s, CH_3 -1') and 2.04 (3H, s, CH_3 -1''), four sp^2 methines at δ_{H} 5.36 (1H, d, $J=10.5$ Hz, H-15), 5.91 (1H, d, $J=10.0$ Hz, H-2), 7.14 (1H, d, $J=10.0$ Hz, H-1), and 7.21 (1H, d, $J=1.5$ Hz,

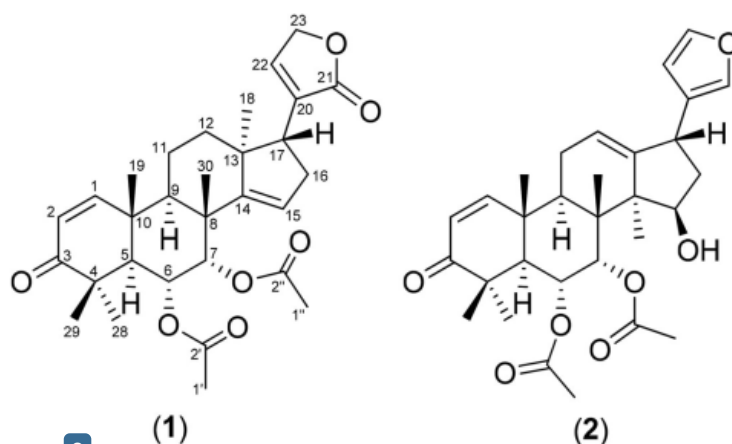


Figure 1. Chemical structures of compounds (1) and (2).

Table 1. ^1H and ^{13}C NMR spectral data for compounds 1 and 2 (500 MHz for ^1H and 125 MHz for ^{13}C).

	Compound 1		Compound 2	
	^{13}C -NMR δ_{C} (mult.)	^1H -NMR δ_{H} [(ΣH ; mult.; J (Hz))]	^{13}C -NMR δ_{C} (mult.)	^1H -NMR δ_{H} [(ΣH ; mult.; J (Hz))]
1	157.3 (d)	7.14 (1H, d, 10.0)	157.3 (d)	7.05 (1H, d, 10.0)
2	126.2 (d)	5.91 (1H, d, 10.0)	125.8 (d)	5.91 (1H, d, 10.0)
3	204.7 (s)	-	204.5 (s)	-
4	40.9 (s)	-	44.6 (s)	-
5	48.0 (d)	2.49 (1H, d, 12.5)	47.4 (d)	2.60 (1H, d, 12.5)
6	69.9 (d)	5.44 (1H, d, 2.5)	70.0 (d)	5.48 (1H, dd, 2.5, 12.5)
7	74.6 (d)	5.42 (1H, d, 2.5)	75.6 (d)	5.73 (1H, d, 2.5)
8	43.2 (s)	-	42.6 (s)	-
9	37.1 (d)	2.26–2.31 (1H, m)	39.7 (d)	2.35–2.37 (1H, m)
10	45.0 (s)	-	41.1 (s)	-
11	16.5 (t)	1.68–1.81 (2H, m)	23.5 (t)	1.96–2.02 (2H, m)
12	32.7 (t)	1.88–1.95 (2H, m)	116.0 (d)	5.30–5.33 (1H, m)
13	47.3 (s)	-	151.5 (s)	-
14	158.1 (s)	-	53.1 (s)	-
15	119.2 (d)	5.36 (1H, d, 3.5)	81.6 (d)	4.16 (1H, br. s)
16	34.1 (t)	2.06–2.08 (1H, m)	42.2 (t)	2.10–2.23 (2H, m)
17	50.9 (d)	2.8 (1H, t, 7.5)	37.7 (d)	3.81 (1H, t, 10.0)
18	20.8 (q)	0.86 (3H, s)	20.2 (q)	1.07 (3H, s)
19	20.5 (q)	1.25 (3H, s)	28.4 (q)	1.29 (3H, s)
20	134.1 (s)	-	127.5 (s)	-
21	174.2 (s)	-	138.9 (s)	7.27 (1H, d, 5.0)
22	146.6 (d)	7.21 (1H, d, 2.5)	109.9 (d)	6.23 (1H, s)
23	70.3 (t)	4.83 (2H, t, 2.5)	143.1 (d)	7.35 (1H, t, 2.0)
28	21.4 (q)	1.18 (3H, s)	20.8 (q)	1.18 (3H, s)
29	31.7 (q)	1.24 (3H, s)	31.4 (q)	1.25 (3H, s)
30	26.9 (q)	1.34 (3H, s)	20.3 (q)	1.33 (3H, s)
2'	170.1 (s)	-	169.9 (s)	-
1'	21.2 (q)	2.00 (3H, s)	21.2 (q)	2.03 (3H, s)
2''	170.3 (s)	-	171.4 (s)	-
1''	21.1 (q)	2.04 (3H, s)	21.6 (q)	2.14 (3H, s)

H-22), three proton signals of oxygenated carbon including one oxygenated methylene at δ_{H} 4.83 (2H, d, $J=2.5$ Hz, H-23), two oxygenated methines at δ_{H} 5.42 (1H, d, $J=2.5$ Hz, H-7) and 5.44 (1H, d, $J=2.5$ Hz, H-6), and several aliphatic signals in the shielded region were also observed.

The ^{13}C NMR and DEPT 135° spectra (Table 1) showed 30 carbon signals and inferred the presence of five quaternary methyl groups that are a characteristic of intact azadirone-type limonoid at δ_{C} 20.8 (C-18), 20.5 (C-19), 51.4 (C-28), 31.7 (C-29), 26.9 (C-30), and two quaternary methyls from two acetoxy groups at δ_{C} 21.2 (C-1') and 21.1 (C-1''). The signals at δ_{C} 457.3 (C-1), 126.2 (C-2), and 204.7 (C-3) in the ^{13}C -NMR spectrum represented the α,β -unsaturated carbonyl moiety at the ring-A. Furthermore, two sp^2 methine carbons at δ_{C} 119.2 (C-15) and 146.6 (C-22), together with two sp^2 quaternary carbons at δ_{C} 158.1 (C-14) and 134.1 (C-20), were also observed as doublets at those positions. One lactone carbonyl at δ_{C} 174.2 (C-21), two ester carbonyls at δ_{C} 170.1 (C-2') and 170.3 (C-2''), and three oxygenated carbon signals at δ_{C} 74.6 (C-7), 69.9 (C-6) and 70.3 (C-23) were observed in the ^{13}C -NMR spectrum. The acetoxy groups were bonded to two oxygenated carbons and the carbon at δ_{C} 70.3 (C-23) was a signal for methylene carbon at the lactone side chain. These functionalities accounted for seven out of the twelve degrees of unsaturation, and five degrees belong to pentacyclic ring of azadirone-type limonoid [15, 16]. The comparison between the NMR spectra of **1** and pentrandicine D isolated from *C. pentandrus* [16] and 23-dehydroxyazadirone isolated from *Azadirachta indica* [17] showed that compound **1** has similar rings A, B, C, and D but with two acetoxy groups and α,β -unsaturated- γ -lactone side chain instead. All the assignments from ^1H , ^{13}C NMR, and DEPT spectra of **1** were supported by ^1H - ^1H COSY, HMQC, and HMBC spectra. The ^1H - ^1H COSY spectrum of compound **1** showed correlations between H₁-H₂, H₅-H₆-H₇, H₉-H₁₁-H₁₂, and H₁₅-H₁₆-H₁₇, and H₂₂-H₂₃. This confirmed the partial structure of intact limonoid azadirone-type skeleton with two acetoxy groups at ring B and α,β -unsaturated- γ -lactone with the lactone carbonyl at the position of C-21. Furthermore, the HMBC experiment was conducted to confirm the position of each functional group of the skeleton (Figure 2), the correlations of H-1 to C-3, C-5, and C-9 supporting the presence of 1-en-3-on moiety at ring A, a typical characteristic of azadirone-type limonoid [18, 19]. It also showed a cross peak from H-15 to C-13, C-16, and C-17 and inferred the position of the double bond between C-14 and C-15. Similarly, the cross peaks of H-22 and H-23 to C-20 and C-21 as well as H-17 to C-20 showed the α,β -unsaturated- γ -lactone with the lactone carbonyl at the position of C-21 and that C-20 connected to C-17.

The relative configuration of **1** was determined by the NOESY experiment (Figure 3). Based on biosynthetic approach, CH₃-30 and CH₃-29 was on the β -orientation while the CH₃-18 was on the α -orientation [20]. The spectrum showed the correlations between H-6, H-7, and H-17 to CH₃-30, indicating both oxygenated protons H-6/H-7 and H-17 were β -orientation and the acetoxy groups, as well as the lactone group, were at the opposite sides at α -orientation. That was supported by the correlations of both methyls of the acetoxy groups and H-22 with CH₃-18 at α -orientation. Furthermore, H-5 and H-9 as protons at asymmetry carbons were determined as α -orientation based on its correlations with CH₃-18. Therefore, compound **1** was elucidated as a new azadirone-type limonoid, named lasiocarpine A.

Lasiocarpine B (**2**) was isolated as a white amorphous solid (MeOH), with a molecular formula of C₃₀H₃₈O₇ following the analysis of HR-TOFMS at m/z 511.2685 [M + H]⁺ with twelve degrees of unsaturation. In addition, the UV spectrum showed

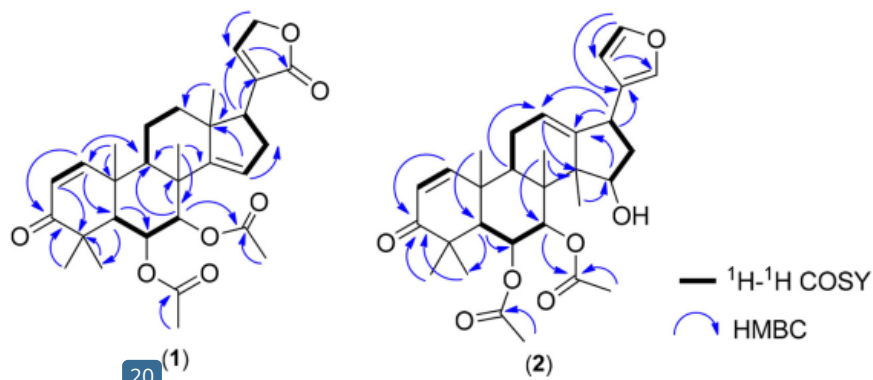


Figure 2. Selected HMBC and ^1H - ^1H COSY correlations.

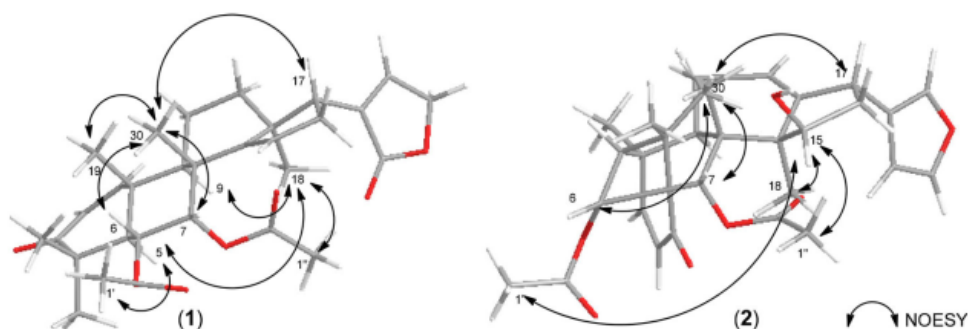


Figure 3. Selected NOESY correlations of compounds 1 and 2.

maximum absorption at 230 nm ($\log \epsilon$ 2.61), indicating the presence of a conjugated system. The IR spectrum were similar with compound 1 besides the presence of hydroxyl group at ν_{\max} 3315 cm^{-1} . In addition, the 1D NMR spectra of 2 (Table 1) were superimposed with compound 4 [21], and showed similar signals except for the presence of sp^3 methine carbon typical of C-17 in compound 2. Additionally, one sp^2 quaternary carbon of the double bond between C-14 and C-17 at 4 was replaced by a sp^2 methine carbon at C-12 of compound 2, and inferred the double bond was between C-12 and C-13. This was supported by the ^1H - ^1H COSY spectrum that showed the correlations between H_9 - H_{11} - H_{12} and H_{15} - H_{16} - H_{35} as well as by HMBC spectrum from the cross peaks between H-9 to C-12, H-12 to C-14 and C-17, H-18 and H-15 to C-13, which confirmed the double bond at positions C-12 and C-13. In addition, the relative configuration was determined by NOESY spectrum, in which both H-7 and H-6 showed correlations with H-30, indicating a similar orientation of these protons. Meanwhile, the hydroxyl group of C-15 at β -orientation was determined based on the correlation of H-15 to both methyls at the acetyl groups and CH_3 -18, and the structure of 2 was inferred as a new azadirone-type limonoid named lasiocarpine B.

The remaining compounds were identified as nimonolactone (3) [17, 22], toonacilactone C (4), and toonacilactone F (5) [21] by comparison of the observed spectroscopy data and previous report.

Table 2. Cytotoxicity of compounds 1–5 against MCF-7 breast cancer cells.

Compounds	IC ₅₀ (μM)
Lasiocarpine A (1)	43.38
Lasiocarpine B (2)	168.04
Nimonolactone (3)	149.15
Toonaciliatone C (4)	827.31
Toonaciliatone F (5)	256.79
Cisplatin*	15.9

Note. *Positive control.

Table 2 showed the cytotoxic activity of compounds 1–5 evaluated against the MCF-7 breast cancer cell line (Table 2) [23]. Based on the IC₅₀ value, compound 1 that contains the γ -lactone and two acetyl groups showed the strongest cytotoxic activity, while compound 3 with one acetyl and γ -lactone group exhibited a significant decrease of the cytotoxic activity. Therefore, the presence of γ -lactone and diacetyl groups have role of cytotoxic activity for these types of limonoids.

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3. Experimental

3.1. General experimental procedures

IR spectra were measured on Thermo Scientific™ Nicolet™ Summit FTIR Spectrometer with DTGS KBr detector and generate with Thermo Scientific™ OMNIC™ Paradigm Software (Thermo Fisher Scientific, Madison, WI, USA), and UV spectra were recorded on TECAN Infinite 200 pro (Männedorf

Switzerland) with methanol p.a as solvent. NMR spectra were measured by JEOL JNM-ECZ500R/S1 spectrometer (Tokyo, Japan) at 500 MHz for ¹H, 125 MHz for ¹³C and TMS as an internal standard, and mass spectra were measured by Waters QTOF-HRTOFMS-XEVOtm mass spectrometer (Waters, Milford, MA, USA). Optical rotations were measured with an ATAGO AP-300 automatic polarimeter (ATAGO, Saitama, Japan), while melting points were measured using the M-565 apparatus (Buchi, Flabourg, Switzerland). Furthermore, column chromatography was conducted on silica gel 60 (70–230 and 230–400 mesh, Merck, Darmstadt, Germany). Thin-layer chromatography was carried out on silica gel 60 F₂₅₄ (Merck, 0.25 mm) and spots were detected under ultraviolet-visible light (λ 254 nm) and followed by spraying with Ehrlich's reagent (*p*-dimethylaminobenzaldehyde in 1:1 hydrochloric acid and ethanol).

3.2. Plant material

Chisocheton lasiocarpus fruits (Miq.) were collected from Bogor Botanical Garden, Bogor, West Java, Indonesia in August 2019. The identification of the plant was conducted by Mr. Harto, the staff of the Bogoriense Herbarium, Bogor, and the voucher specimen (VII. G. 168) was deposited at the herbarium.

3.3. Extraction and isolation

The dried fruit (1.2 kg) was extracted using methanol for 6 days and the solvent was evaporated to produce concentrated methanol extract (270 g). Furthermore, the extract was fractionated successively with *n*-hexane, ethyl acetate, and *n*-butanol followed by solvent removal to obtain concentrated *n*-hexane (36.5 g), ethyl acetate (28.1 g), and *n*-butanol (23.6) extracts.

The ethyl acetate extract was separated by vacuum liquid chromatography on silica gel 60 using a gradient of *n*-hexane and ethyl acetate to yield seven fractions (E.A–E.G). Meanwhile, fraction E.B (0.4 g) was chromatographed on silica gel using dichloromethane and ethyl acetate (8:2) solvent to obtain four subfractions (E.B1–E.B4). Subfraction E.B1 (45 mg) was chromatographed on ODS eluted with methanol: water (8:2) to produce compound 1 (5 mg). Subfraction E.B2 (50 mg) was chromatographed on ODS with methanol: acetonitrile: water (4:3:3) to yield compound 2 (3.5 mg). Similarly, subfraction E.B3 (55 mg) was chromatographed on ODS using an isocratic solvent of methanol: acetonitrile: water (4:4:2) to yield compound 3 (4 mg). Fraction E.C (2.4 g) was chromatographed on silica gel using gradient solvent of *n*-hexane and ethyl acetate to yield five subfractions (E.C1–E.C5). Subfraction E.C3 (0.9 g) was chromatographed on silica gel using *n*-hexane: dichloromethane: ethyl acetate (5:3:2) to yield three fractions (E.C3A–E.C3C). Also, subfraction E.C3B (0.2 mg) was chromatographed on ODS using an isocratic solvent of methanol: water (7:3) to yield compounds 4 (27 mg) and 5 (31 mg).

3.3.1. Lasiocarpine A (1)

Colorless amorphous solid; m.p. degraded under its melting point; $[\alpha]_D^{28.2} + 12.1$ (c 0.51, CHCl₃); UV (MeOH) λ_{\max} (log ϵ): 230 nm (2.62); IR (KBr) ν_{\max} 2917, 2848, 1729, 1676, 1447, and 1084 cm⁻¹; ¹H-NMR spectral data (CDCl₃, 500 MHz) see Table 1; ¹³C-NMR spectral data (CDCl₃, 125 MHz), see Table 1; HR-TOFMS (positive ion mode): *m/z* 511.2691 [M + H]⁺ (calcd for C₃₀H₃₉O₇, 511.2696).

3.3.2. Lasiocarpine B (2)

White amorphous solid; m.p. degraded under its melting point; $[\alpha]_D^{28.3} + 17.6$ (c 0.23, CHCl₃); UV (MeOH) λ_{\max} (log ϵ): 230 nm (2.61); IR (KBr) ν_{\max} 3315, 2955, 1731, 1668, and 1042 cm⁻¹; ¹H-NMR spectral data (CDCl₃, 500 MHz) see Table 1; ¹³C-NMR spectral data (CDCl₃, 125 MHz), see Table 1; HR-TOFMS (positive ion mode): *m/z* 511.2685 [M + H]⁺ (calcd for C₃₀H₃₉O₇, 511.2696).

3.4. Determination of cytotoxic activity

The cytotoxic activity was determined by the same PrestoBlue® assay as previously reported by Nurlelasari et al., 2021 [23].

Acknowledgements

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Disclosure statement

No potential conflict of interest was reported by the authors.

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